

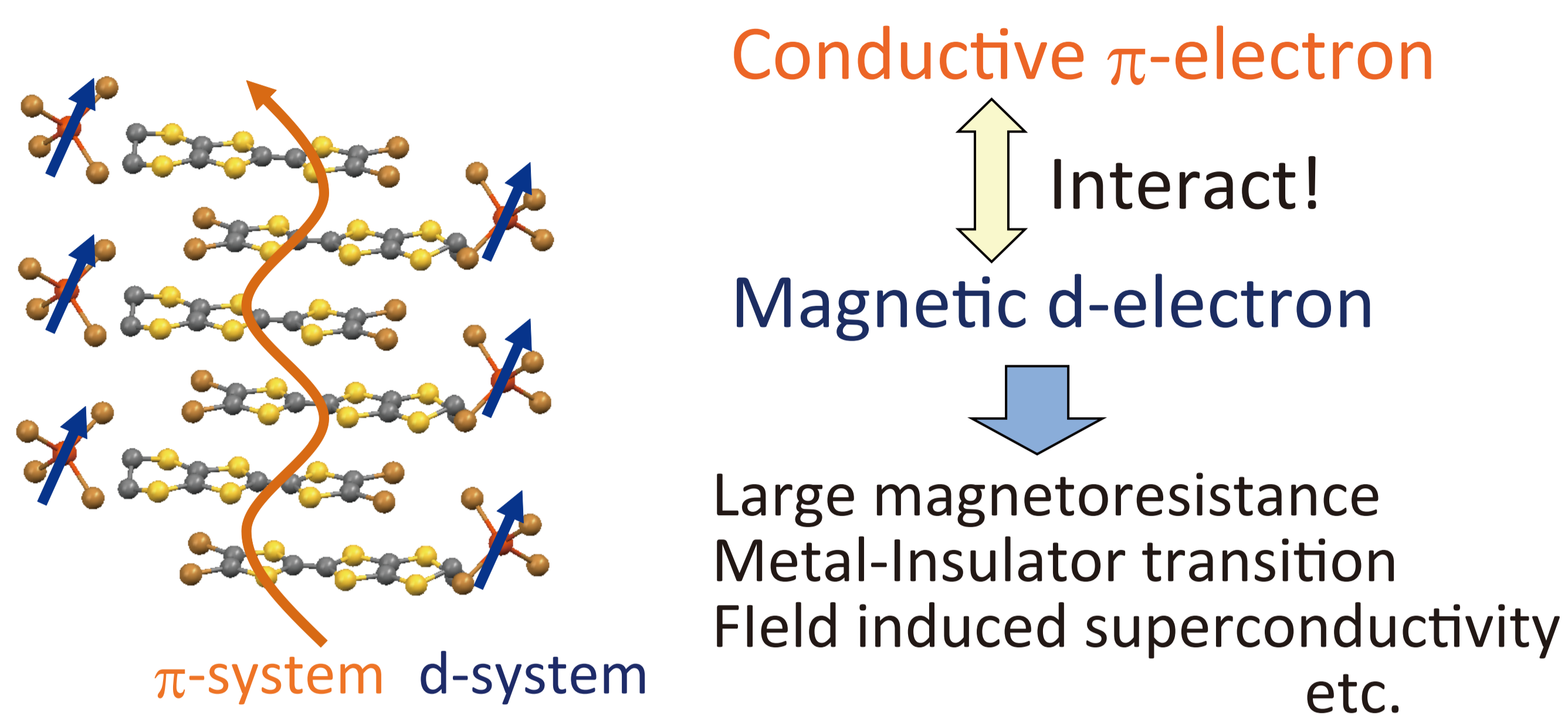


# Chromium Acetylide Based Magnetic Materials

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## 1. Introduction: $\pi$ -d system



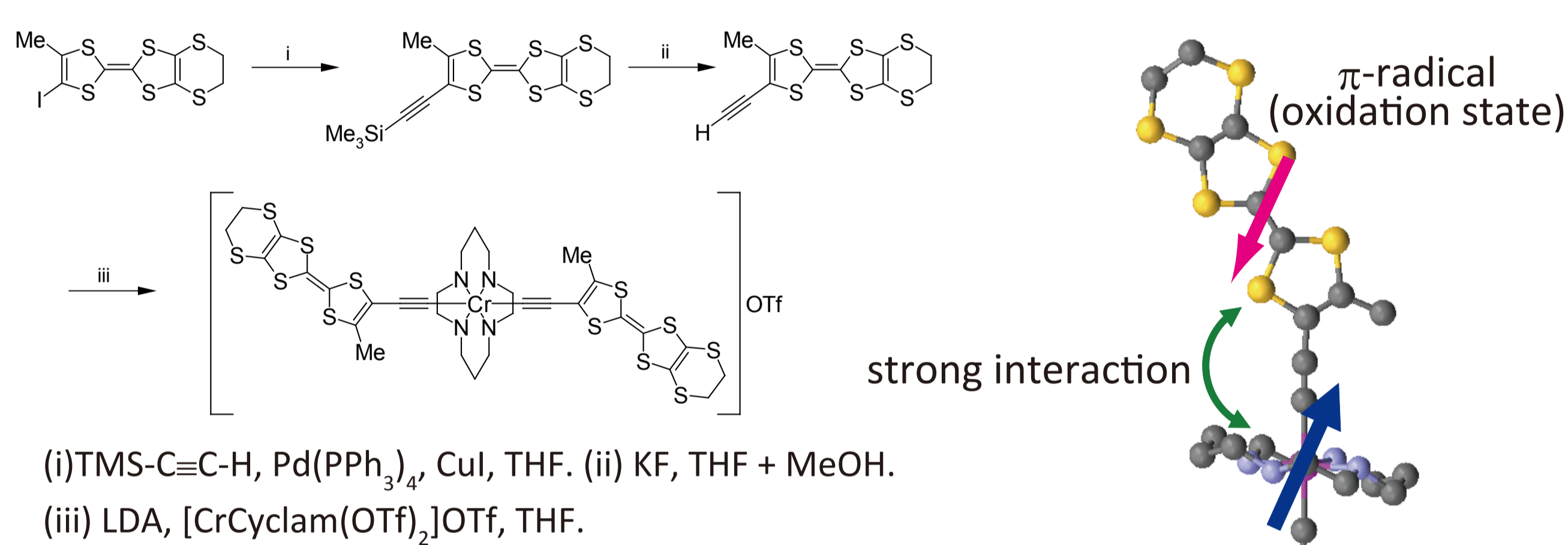
However, most of the  $\pi$ -d interaction is quite weak...

Why weak?

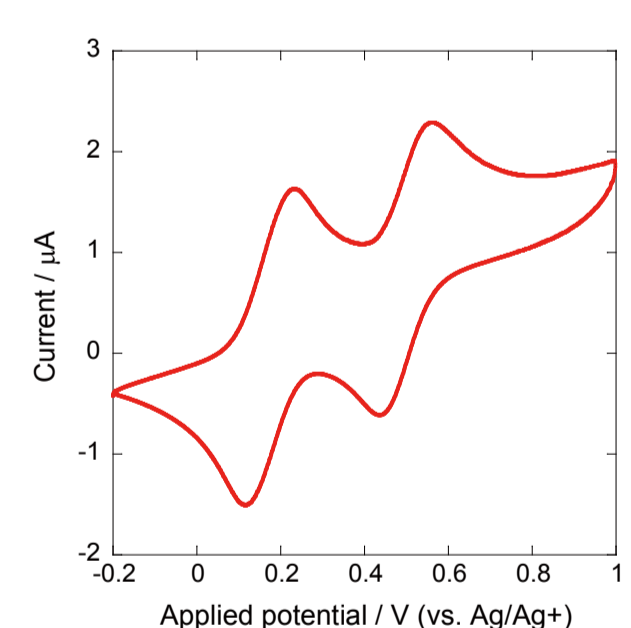
Magnetic d-electrons and conductive  $\pi$ -electrons are placed on distant molecules.

Integrate the  $\pi$ - and d-electrons into a molecule!

## 2. Chromium-Acetylide-TTF complex



- Stable complex
- Redox active TTF-type ligand
- Strong intra-molecular interaction (oxidation state)



- Two reversible oxidation peaks  
→ 1<sup>st</sup> and 2<sup>nd</sup> oxidation of TTF

### Crystal growth of radical cation salts:

Standard galvanostatic oxidation in PhCl + MeCN (1:1) solution of Bu<sub>4</sub>NClO<sub>4</sub> or Bu<sub>4</sub>NBF<sub>4</sub>.

## 5. Conclusion

The first step toward the strongly interacted  $\pi$ -d system

We successfully synthesized a new complex with TTF ligand.

In the crystals obtained by electrochemical oxidation;

TTF backbones form mixed valence dimer ( $S = 1/2$ )

The complex forms 1D ferrimagnetic chain with

strong intra-chain interaction ( $2J \sim -30$  K)

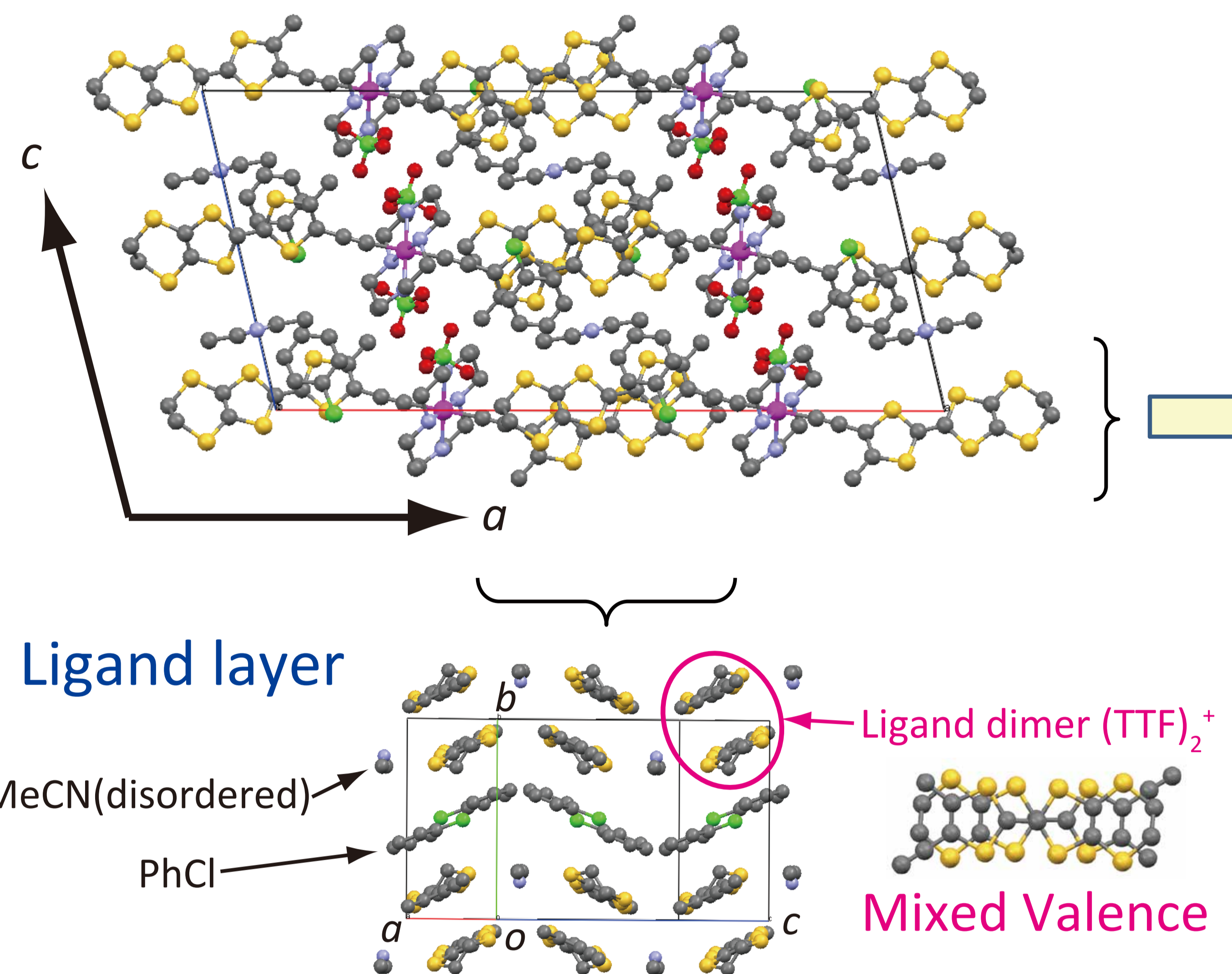
Weakferromagnetic transition  $T_N = 23$  K

The strong intra-molecular  $\pi$ -d interaction of Chromium-Acetylide-TTF type complex is promising for constructing novel  $\pi$ -d systems.

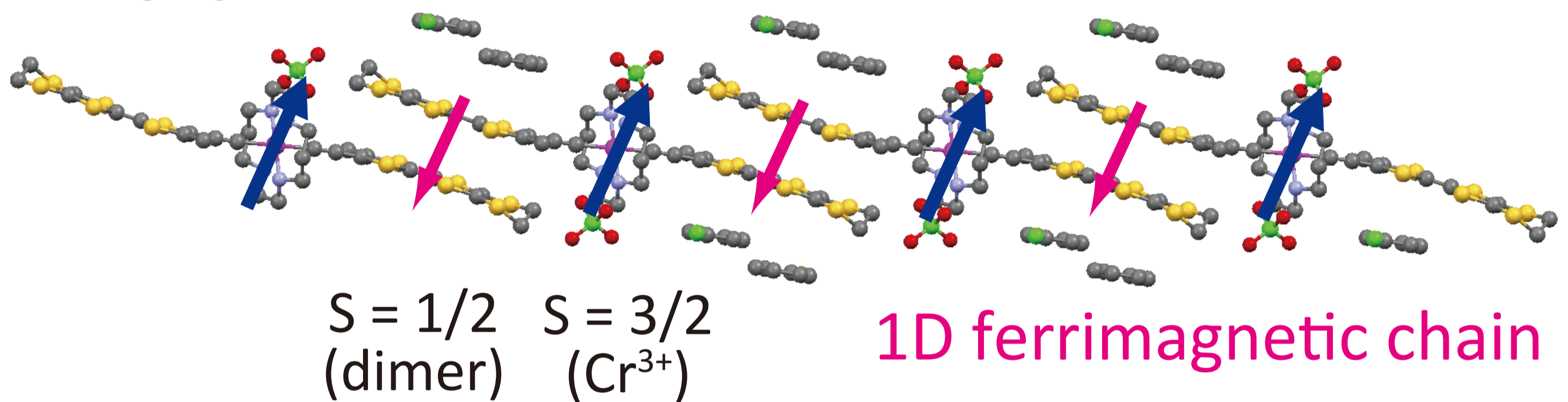
## 3. Crystal structure of



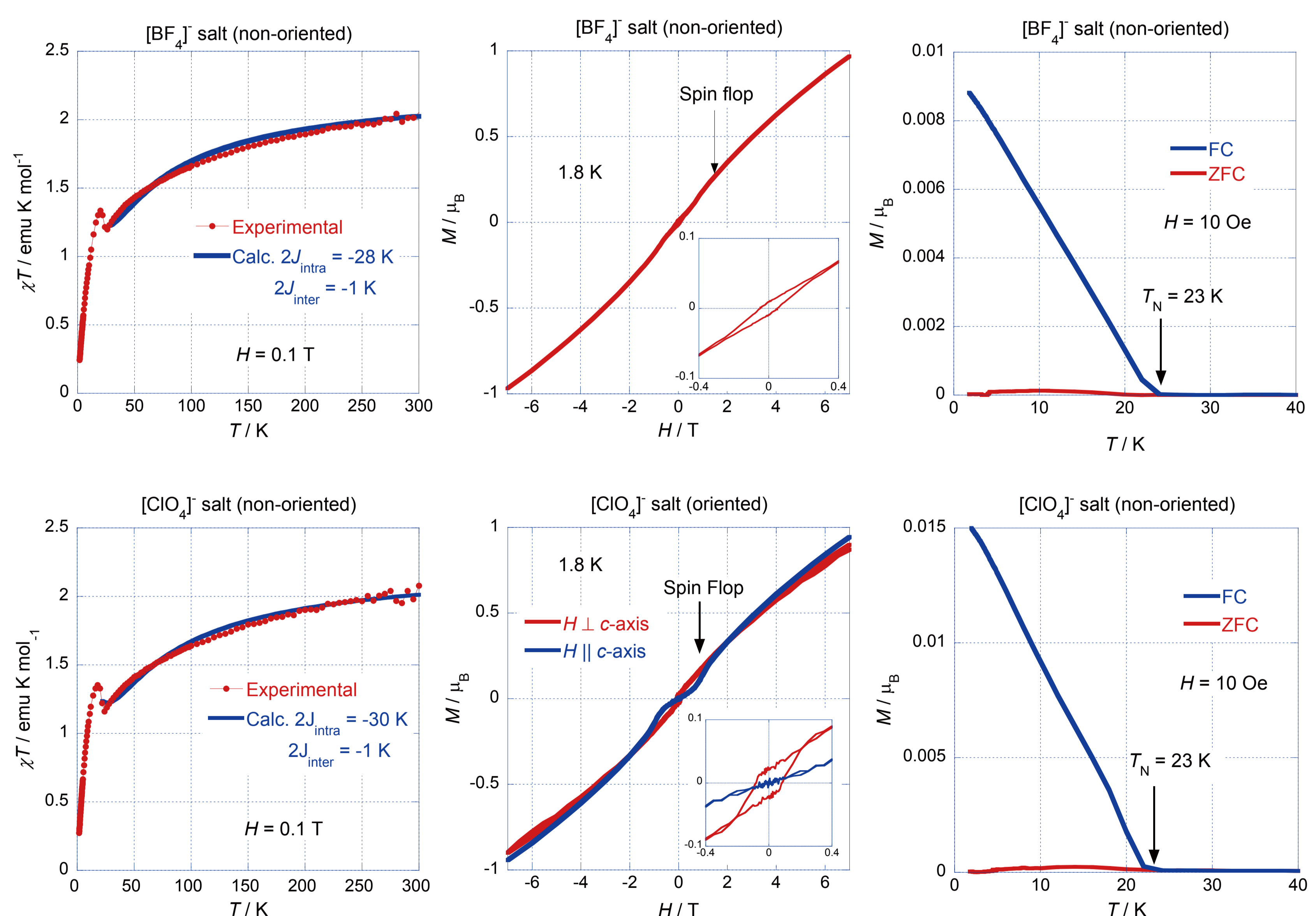
(same crystal structure regardless of anion X = [ClO<sub>4</sub>]<sup>-</sup>, [BF<sub>4</sub>]<sup>-</sup>)



### 1D chain



## 4. Magnetic properties



Strong intra-chain interaction  $2J = -28$  and  $-30$  K

High weakferromagnetic transition temperature  $T_N = 23$  K

Remanent magnetization:  $0.009 \mu_B$  for [BF<sub>4</sub>]<sup>-</sup> salt (non-oriented)

$0.023 \mu_B$  for [ClO<sub>4</sub>]<sup>-</sup> salt (oriented)

Coercive force: 50 mT for [BF<sub>4</sub>]<sup>-</sup> salt (non-oriented)

85 mT for [ClO<sub>4</sub>]<sup>-</sup> salt (oriented)

Origin of the weakferromagnetism:

Single ion anisotropy and / or Dzyaloshinsky-Moriya interaction